

Caking Mechanisms of Passion Fruit Powder During Storage

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Dry food is generally susceptible to caking during storage. Therefore, the purpose of this study is to analyse the influence of temperature and relative humidity on mechanisms contributing to the caking of passion fruit powder during storage. The research was conducted by storing passion fruit powder in desiccators at various points of temperature (10, 20, and 30°C) and relative humidity (32.8%, 52.8%, and 75.2%). The variations in storage RH were produced by using saturated salt solutions put under the desiccators. The parameters were the evaluated extent of passion fruit powder caking during storage. The results showed that variation in storage temperature did not significantly influence caking because the storage temperature used in this study was still lower than the glass transition temperature of the samples. Water molecules in this condition become unable to carry out chemical activities that cause caking. However, RH variation has a significant influence on the caking of passion fruit powder. The high moisture content at a storage RH of 52.5 and 75.2% could trigger caking.

Keywords: *Passion fruit, Powder, Caking, Temperature, Relative Humidity*

Introduction

The passion fruits (*Passiflora* spp.), including the Passifloraceae family, consist of more than 500 species. Among them, the well-known genus is *Passiflora* (Pereira, Correa, & Oliveira, 2015). Spanish missionaries discovered *Passiflora* species in South America to be medicinal plants (Ozarowski & Thiem, 2013).

At present, *Passiflora* species are widely used in traditional medicine throughout the world, mainly as sedatives and anxiolytics, which are also exploited by the food, pharmacology, and cosmetics industries (Marques, et al., 2016). The most popular species in Indonesia is *Passiflora edulis* Sims because it can be consumed either fresh or processed (Ansar, Suhargo, & Rahardjo, 2004). Consumers prefer it because it contains nutrients such as beta-carotene,

potassium, fibre, and vitamin C. Some believe it reduces blood pressure because it is beneficial to health (Lopez-Vargas, Fernandez-Lopez, Perez-Alvarez, & Viuda-Martos, 2013; Ramaiya, et al., 2013).

Passion fruit can be processed into powder as raw material for instant drinks, baby food ingredients, ice cream, bread, and snacks. However, the problem of its powder is hygroscopic, so it cakes easily (Alakali & Satimehin, 2009; Ansar, Rahardjo, Noor, & Rochmadi, 2011).

In Indonesia, powdered fruit products have not been developed widely (Hariadi, 2013; Ansar, Nazaruddin, & Azis, 2019). In other countries, fruit processing into powders has been developed with satisfactory results (Matsui et al., 2010; Pack, Kim, Pack, & Kim, 2012). The problem that always occurs in passion fruit powder is that the particles easily cake and change colour, becoming brown during storage.

The physical change caused by caking has a negative impact on consumer judgment, both in terms of taste and nutrition (Akhtar, Abbasi, & Hussain, 2010; Liu, et al., 2008). Therefore, caking can be used as an indicator of the expired shelf life of the powdered product. Caking can be overcome by anti-caking. However, consuming a lot of food containing preservatives can affect one's health (Kandandapani, Balaraman, & Ahamed, 2015; Lima, Vuolo, Batista, & Drag, 2016).

The caking process in powdered products has not been publicised widely. Therefore, it is very important to analyse the causes of caking in passion fruit powder. Based on these arguments, the purpose of this study is to analyse the influence of temperature and relative humidity on the caking of passion fruit powder during storage.

Methods

Materials

The raw materials used were passion fruit collected from a farmer in Malino, South Sulawesi, Indonesia, to be produced into passion fruit powder. The saturated salt solutions used to control relative humidity (RH) in the desiccator were $MgCl_2$ for RH 32.8%, $Mg(NO_3)_2$ for RH 52.8%, and NaCl for RH 75.2%. The equipment used in these experiments, among others, were pulpier sieve, a vacuum freeze dryer, homogeniser, freezer, and oven dryer.

Preparation of the Passion Fruit Powder

The research procedure was carried out in several steps, including drying passion fruit juice using a vacuum dryer to produce passion fruit powder with a moisture content of 14% (Moghadam & Sani, 2014). The passion fruit powder was stored in the desiccators with temperature variations at 10, 20, and 30°C. RH was 32.8%, 52.8%, and 75.2%. Storage with RH variations was carried out using saturated salt solutions (Ansar, Nazaruddin, Alamsyah, & Azis, 2018). Before being stored, the water content of passion fruit powder was determined based on the standard analysis method (AOAC, 2016).

Mathematical Model Development

The caking parameter of the passion fruit powder during storage follows the model of the first order kinetic reactions that can be expressed by the equation (Ansar, Rahardjo, Noor, & Rochmadi, 2011):

$$P = P_0 \exp(k_p \cdot \theta) \quad (1)$$

with P = caking parameter of the passion fruit powder, P_0 = initial caking value of the passion fruit powder, k_p = rate constant of caking during storage, and θ = storage duration.

If the value of k_p was a function of both temperatures (T) and relative humidity (RH), then k_p can be written as (Ansar, Rahardjo, Noor, & Rochmadi, 2011):

$$k_p = f(T) - f(RH) \quad (2)$$

Value k_T as a function of temperature at RH constant follows the Arrhenius equation:

$$k_T = A \exp(-Ea / RT) \quad (3)$$

While the value of k_{RH} as RH function at constant temperature is:

$$k_{RH} = B(RH)^{k_R} \quad (4)$$

Substitution of equations (3) and (4) into equation (2) results in a caking rate prediction equation where k_{RH} and k_T are combined into a single constant C :

$$k_p = C(RH)^{k_R} \exp(-Ea / RT) \quad (5)$$

Thus, equation (1) can be rewritten as:

$$P = P_0 \exp[C(RH)^{k_R} \exp(-Ea / RT)]\theta \quad (6)$$

Caking Observation

Caking observation of the passion fruit powder was conducted every week for 3 months. The passion fruit powder was called caked if the particles could not filter through sieve 10 mesh (Suyitno, 1995). Their percentage was analysed by a procedure such as was done with milk powders (Freeman, Brockbank, & Armstrong, 2015).

Statistical Analysis

The experimental data was analysed using regression analysis. A rate change constant of caking in the passion fruit powder (k_p) was a function of temperature. RH was calculated by plotting $\ln k_p$ vs inverse of absolute temperature $1/T$. The line slope obtained from the equation of a straight line is E_a/R , and a line intersection k_p value was used to predict the caking rate (Ansar, Rahardjo, Noor, & Rochmadi, 2011).

Results and Discussion

The passion fruit powder had a soft texture and a light yellow colour (Figure 1).

Figure 1. The passion fruit powder before (A) and after (B) storage

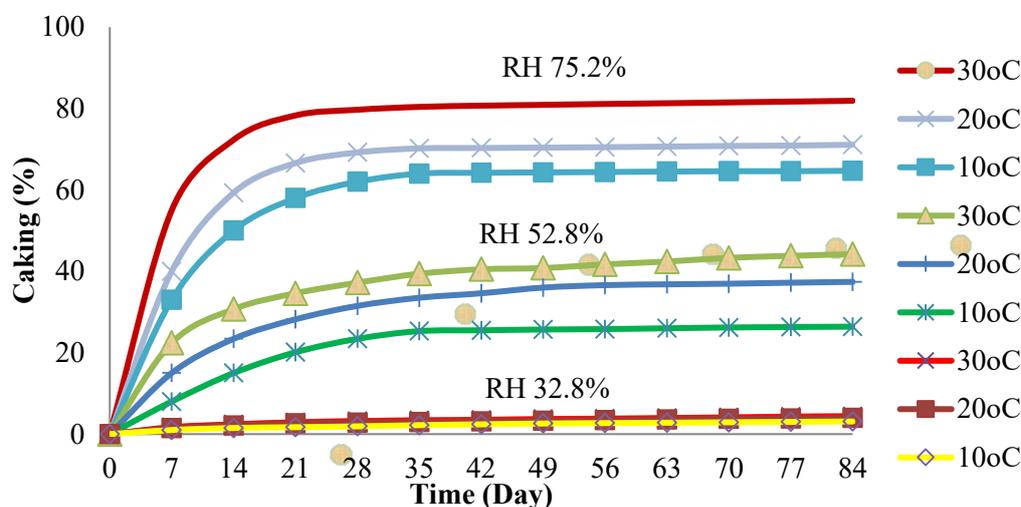


Although the aroma was not as strong as the aroma of fresh passion fruit, the taste remained the same as fresh fruit. The results of measuring the water content of passion fruit powder ranged from 3.07 to 3.66% db. These results are not significantly different from other dry food products, such as pure pumpkin powder 3.99% (Dirim & Caliskan, 2012). Coconut sugar powder was 1.92-3.03% (A-sun, Thumthanaruk, Lekhavat, & Jumnonng, 2016), while arenga pinnata sugar was 4.11% (Choong, Anzian, Che Wan Sapawi, & Moer Hussin, 2016).

Caking was an indicator of changes in the quality of passion fruit powder, which was most easily observed during storage. Caking occurred because there was a layer of adhesive on the surface of the particles that could solve the sugar compounds and form a sticky mass (Al-Hashemi & Al-Amoudi, 2018). During storage, there was also a process of absorption of steam on the surface of the sample, which affected caking. Caking occurs because there is an interaction between material and air humidity through a chemical process (Fitzpatrick, Podczec, & Weigl, 2010).

During storage, there was an increase in the percentage of caking in the sample (Figure 2). At 32.8% RH storage, with variations in temperatures of 10, 20, and 30°C for three months, it was found that passion fruit powder had not clumped. When passion fruit powder was stored at RH 52.5 and 75.2% with 3 temperature variations of 10, 20 and 30°C, caking occurred. This data shows that RH storage significantly affects the caking of passion fruit powder. The same case has been reported by Suyitno (1995): the higher the RH in storage, the more particles that cake. In his research, he showed that the percentage of acid powder caking reached 100% at 75% RH storage for three months.

Figure 2. Graphs of caking of passion fruit powder during storage at temperature and RH variations

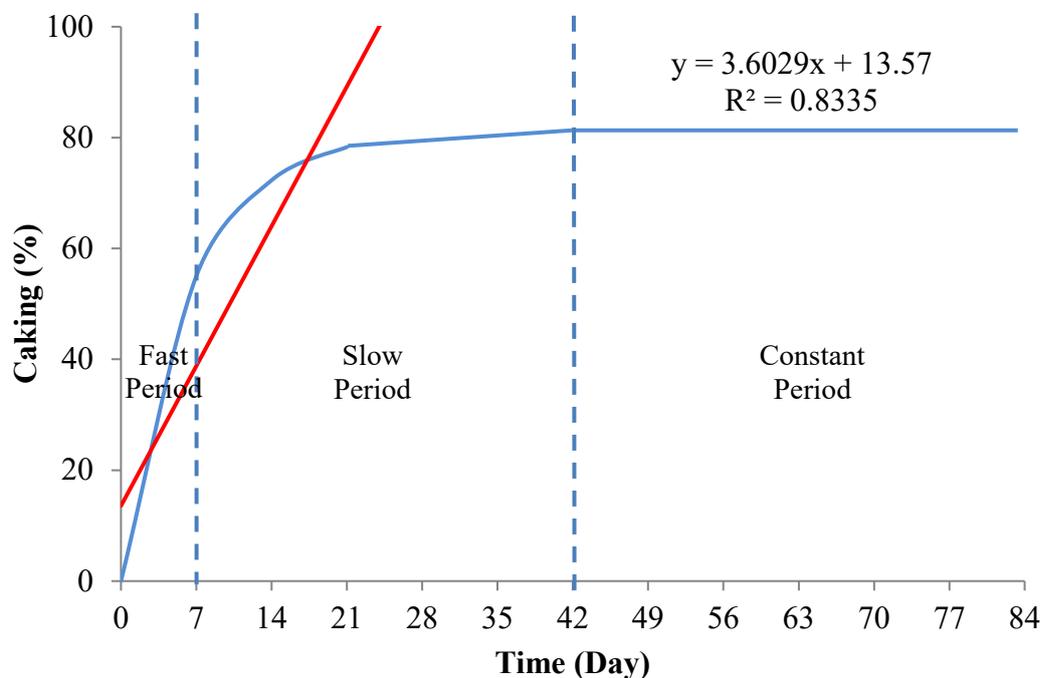


However, variations in storage temperature of 10, 20, and 30°C do not significantly influence the caking of passion fruit powder (Figure 2). This was estimated because the storage temperature used in this study was still below the glass transition temperature of passion fruit powder of 31°C. The same research has been reported by Roos, Karel, & Kokini (1996) and Janzanti & Monteiro (2014), who found that storing materials below the glass transition temperature can maintain product shelf life because there is no change in molecular mobility that can affect the rate of chemical reactions, enzymatic processes, and material purity.

Khalloufi, El-Maslouhi, & Ratti (2006) have also described that glass transition temperatures can be used as standard storage temperatures for powdered products.

The trigger for caking of passion fruit powder was also influenced by RH storage. The caking model of passion fruit powder during storage could be divided into three stages, namely fast, slow, and constant (Figure 3). During storage from the 1st 7th day, the caking curve tends to increase and be linear. This is a very fast caking period. On the 21st days of storage, the caking decreases (slow period), and by the 42nd day, all particles have clumped 100% (constant period).

Figure 3. The caking trend of the passion fruit powder during storage



Increased caking during storage was closely related to high humidity (52.8% and 75.2%). At low RH storage (32.8%), the moisture content was saturated, so there was no caking. The high moisture vapor contents at RH 52.8% and 75.2% can cause caking. Mariela (2014) has reported the same study results. Agudelo-Laverde, Florencia, & Pilar (2014) and Seifu, Tola, Mohammed, & Astatkie (2018) indicate that during storage with RH above 60%, there was water vapor that could function as an adhesive layer on the surface of the particles that could cause caking.

The largest percentage of caking occurred in a fast period, reaching 80%. If compared with a slow and constant period, caking that happened in a fast period could be a guideline for practical actions, such as packaging, storage, and distribution.

The mechanism of the caking process during storage began with the process of absorption of moisture on the surface of the particles, which then formed a thin layer. This thin layer caused bonding on the surface of the particle. As storage time progresses, the stickier particles then form a mass.

The results of the evaluation of k_T and k_{RH} constants, as functions of temperature and RH variations, are shown in Table 1. The table shows that with longer storage and higher temperature, the k_T value also increased. The same thing also happens in RH variation: the longer storage and the higher RH, the higher k_{RH} .

Table 1. The evaluation results of k_T and k_{RH} constants as functions of temperature and RH variations

| No. | Treatments | | k_T | k_{RH} |
|-----|------------------|--------|--------|----------|
| | Temperature (°C) | RH (%) | | |
| 1 | 10 | 32.8 | 0.0043 | 0.0012 |
| 2 | 20 | 32.8 | 0.0046 | 0.0021 |
| 3 | 30 | 32.8 | 0.0047 | 0.0032 |
| 4 | 10 | 52.8 | 0.0024 | 0.0016 |
| 5 | 20 | 52.8 | 0.0029 | 0.0026 |
| 6 | 30 | 52.8 | 0.0036 | 0.0036 |
| 7 | 10 | 75.2 | 0.0012 | 0.0018 |
| 8 | 20 | 75.2 | 0.0023 | 0.0029 |
| 9 | 30 | 75.2 | 0.0027 | 0.0045 |

The influence of temperature on the caking rate constant (k_T) during storage can be known from the determination coefficient value (R^2) of 0.941. This value indicates there was a close relationship between temperature and k_T during storage.

The plot between $\ln k_T$ and $1/T$ was obtained with a line equation whose slope was the E_a/R -value and the line intersection was the k_T value, written as:

$$\ln k_T = -6.4628(1/T) + 0.0241 \quad (7)$$

Equation (7) obtained the value of E_a at 72,237 kJ/mol, and k_P was 4.172E+10. These values indicate that the energy required for the caking activation process of the passion fruit powder during storage was very large.

The influence of RH on the caking rate constant (k_{RH}) of the passion fruit powder during storage could be known from the determination coefficient value (R^2), which was 0.965. This value showed there was a close relationship between RH and k_{RH} during storage.

A plot between $\ln k_{RH}$ and storage RH obtained a line equation:

$$\ln k_{RH} = 3 \exp - 05(RH) + 0.0021 \quad (8)$$

The values of k_T and k_{RH} showed the caking rate of the passion fruit powder during storage. The larger the values of k_T and k_{RH} , the faster the caking rate. The results of this study are in line with the opinion's of Shimoni & Labuza (2000): large constant values indicate changes in high food quality.

Conclusion

Storage temperature variation did not significantly influence the caking because the storage temperatures used in this research were still lower than the glass transition temperature of passion fruit powder. This condition caused the water molecules to be unable to perform such chemical activity that caused caking. However, storage RH variation significantly influenced the caking of the passion fruit powder. The high moisture content at a storage RH of 52.5% or 75.2% could trigger caking. The mechanisms of the caking process during storage begin with the process of absorption of moisture on the surface of the particles, forming a thin layer. This thin layer causes bonding on the surface of the particles. As storage time progresses, the stickier particles then form a mass. Therefore, storage at a temperature of 10°C and RH 32.8% are recommended to prevent caking during storage.

Nomenclature

- θ : Storage duration
- E_a : Energy activation
- k_p : Rate constant of caking
- k_{RH} : Relative humidity constant
- k_T : Temperature constant
- P : Caking parameter
- P_0 : Initial caking value
- RH : Relative humidity (%)
- R^2 : Coefficient of determination
- T : Temperature (°C)

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